

FULL PAPER

Acid-Induced Rearrangement of Cycloadducts from Cyclopropenecarboxylates and 1,3-Diarylisobenzofurans

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Treatment of several *Diels–Alder* adducts of cyclopropenecarboxylates and 1,3-diarylisobenzofurans with a strong acid triggers a skeletal rearrangement resulting in 4,8b-dihydro-3a*H*-indeno[1,2-*b*]furans.

Keywords: Cyclopropenecarboxylates, 1,3-Diarylisobenzofurans, Rearrangements, Indeno[1,2-*b*]furans, *Diels–Alder* adduct.

Introduction

Natural and synthetic products containing an indeno[1,2-*b*]furan framework display a variety of pharmacological effects including antibacterial [1][2] and enzymatic [1] activities. Indenofuran derivatives are also used as a stimulant for the germination of seeds of parasitic weeds belonging to the genera *Striga* and *Orobanche* [3 – 6]. There has been interest in the synthesis of indeno[1,2-*b*]furan ring systems and several multistage approaches have been reported [4][7][8]. Several approaches to the synthesis of the indeno[1,2-*b*]furan ring system have been developed, including methods such as reaction of ninhydrin with diketones [1][2][9][10], titanocene-catalyzed cyclocarbonylation of *o*-allyl aryl ketones [11], electrochemical reduction of 2,2-dibromo-1,3-diketones in the presence of indene [12], oxidative addition of 1,3-dicarbonyl compounds to indene [13][14]. Bhuyan and co-workers have reported a novel synthetic method for the preparation of dihydroindeno[1,2-*b*]furans *via* one-pot three-component reaction of 1,3-indanedione, aromatic aldehyde, and a pyridinium ylide in the presence of Et₃N under MW irradiation [15]. In 2014, Yang reported the first example of one-pot three-component condensation reactions of 1,3-indandione, aromatic aldehyde, and cyclohexyl isocyanide with the formation of indeno[1,2-*b*]furan derivatives in good yields [16].

Results and Discussion

It was discovered that the *Diels–Alder* adducts of 1,3-diarylisobenzofurans and cyclopropenecarboxylates on treatment with a strong acid undergo a cascade of cationic rearrangements leading to compounds with an indeno

[1,2-*b*]furan skeleton. In this article, we present the synthesis, structural elucidation, and a mechanistic proposal for this rearrangement. Previously, acid-induced rearrangements of 1,3-diphenylisobenzofuran adducts with azirines [17 – 19], maleimides [20 – 23], (*E*)-6,6-dimethoxy-3-hexen-2-one [24], dimethyl maleate [23], cyclooctene [23], cyclopropenes [25 – 27], and *N*-aryl itaconimides [28] have been studied. Cava and Narasimhan reported that the [4+2] adduct from 1,3-diphenylisobenzofuran and 1,2,3-triphenylcyclopropene by treatment with hydrochloric acid underwent the fragmentation affording the 1,2,3,4-tetraphenylnaphthalene and benzaldehyde [26]. Similar results were obtained by treatment of the *Diels–Alder* adduct from 1,3-diphenylisobenzofuran and 1,2,3-triphenylcyclopropene with MeSO₃H in CH₂Cl₂ at ambient temperature (*our data*).

For this study, we prepared several *Diels–Alder* adducts **3** from 1,3-diarylisobenzofurans **2a**, **2b** and cyclopropenecarboxylates **1a**, **1b**. The reactions of compounds **2a** with **1a**, **1b** and **2b** with **1b** in benzene at 60° resulted in the *exo*-adducts **3a** – **3c** (Table). It should be noted that according to the ¹H-NMR data, the reaction mixtures contained only the starting compounds and *exo*-adducts **3a** – **3c**. The *endo*-isomer could not be detected. The reaction products were isolated by column chromatography (CC) over silica using a mixture of hexane/AcOEt as the eluent. The structures of adducts **3a** – **3c** were characterized from their ¹H- and ¹³C-NMR, MS, and IR spectra. The ¹H-NMR spectra of adducts **3a**, **3b**, and **3c** exhibit signals at δ(H) 3.60, 3.83, and 3.77, respectively, belonging to the cyclopropyl H-atoms. The values of the chemical shifts for cyclopropyl H-atoms indicate their *syn*-position with respect to the O-atom [29 – 31].

The rearrangement of compounds **3** was found in the presence of acids. Of the acids (MeSO₃H, H₂SO₄, HCl,

Table. Synthesis of *Diels–Alder* adducts **3a**–**3c**

Entry	R ¹	R ²	Time [h]	Yield of 3 [%] ^{a)}
1	Me (1a)	Ph (2a)	1 h	83 (3a)
2	Ph (1b)	Ph (2a)	3 h	80 (3b)
3	Ph (1b)	4-BrC ₆ H ₄ (2b)	3 h	52 (3c)

^{a)} Yield of isolated product.

BF₃·OEt₂) and solvents (CH₂Cl₂, benzene, toluene, THF, Et₂O) screened, the combination of MeSO₃H and CH₂Cl₂ produced the best results for these reactions [20].

In our initial studies, the reaction of **3a** with MeSO₃H was investigated (*Scheme 1*). Compound **3a** was reacted with MeSO₃H (6 equiv.) in CH₂Cl₂ at ambient temperature. After stirring for 6 h and extractive workup, the thin-layer chromatography (TLC) analysis indicated the formation of several compounds. Only methyl (3a*SR*,4*RS*,8*bRS*)-4,8*b*-dihydro-2-methyl-3a,4,8*b*-triphenyl-3a*H*-indeno[1,2-*b*]furan-3-carboxylate (**4a**) and methyl 3-oxo-2-(1,1,3-triphenyl-1*H*-inden-2-yl)butanoate (**5**) were isolated from the reaction mixture by a CC in 35% and 15% yields, respectively; no other products were isolated in pure form (*Scheme 1*). After reducing the reaction time to 3 h, we could isolate product **4a** in 51% yield. It should be noted that using one or two equivalents of MeSO₃H in the reaction mixture after stirring for 6 h was observed the presence of starting *Diels–Alder* adduct **3a**. According to TLC data, using 1–3 equivalents of the MeSO₃H leads to more complex reaction mixtures. The compositions and structures of the products **4a** and **5** were established by elemental and spectroscopic analyses. The ¹H-NMR spectrum of indeno[1,2-*b*]furan **4a** exhibits a signal for the CH H-atoms at the C(4) atom at δ(H) 5.7, and a signal for the Me group at the C(2) atom at δ(H) 2.4. The fused tricyclic structure and relative configuration of product **4a** were confirmed by a single-crystal X-ray diffraction (*Fig. 1*). The ¹H-NMR spectrum of indene **5** shows *singlet* signals at δ(H) 4.1 and 2.2 belonging to the H-atom in α-position of the ester group and the MeCO

group, respectively. The ¹³C-NMR spectrum of compound **5** exhibits a signal for the CO group at δ(C) 208, signals for the Me groups at δ(C) 52 (CO₂Me), and 31 (MeCO), a signal for the CH C-atom at δ(C) 55 and a signal for C(1) at δ(C) 65. We proposed that indene **5** was formed from indenofuran **4a** under acidic conditions, and when **4a** was treated with MeSO₃H in CH₂Cl₂ the formation of indene **5** was observed.

For the next step in our study, we treated *Diels–Alder* adduct **3b** with MeSO₃H (*Scheme 2*). The reaction proceeded at room temperature producing indeno[1,2-*b*]furan

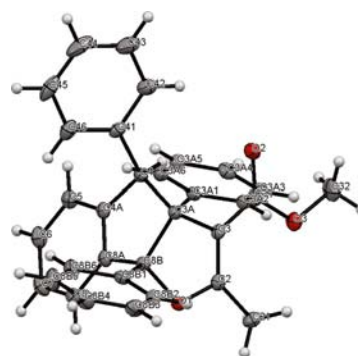
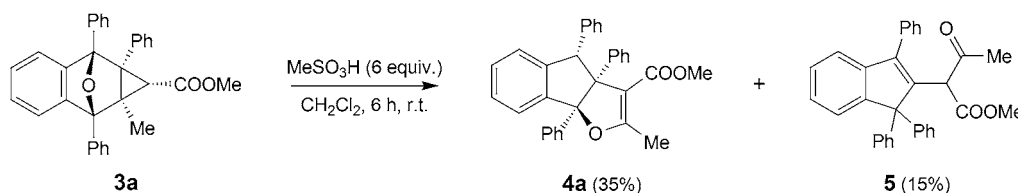


Fig. 1. Molecular structure (ORTEP 50% probability level) of **4a**. Selected interatomic distances [Å] and Angles [°]: O(1)–C(2) 1.3578 (13), O(1)–C(8*b*) 1.4720(12), C(2)–C(3) 1.3470(15), C(3)–C(3a) 1.5265 (14), C(3a)–C(4) 1.5924(14), C(3a)–C(8*b*) 1.5990(14), C(4)–C(4a) 1.5094(14), C(4a)–C(8*a*) 1.3835(15), C(8*a*)–C(8*b*) 1.5059(14), C(2)–O(1)–C(8*b*) 108.35(8), C(3)–C(2)–O(1) 113.25(9), C(3)–C(3a)–C(4) 108.75(8), C(3)–C(3a)–C(8*b*) 98.64(8).

Scheme 1. Reaction of *Diels–Alder* adduct **3a** with MeSO₃H.

4b in 37% yield. Although the TLC analysis indicated the formation of several compounds, isolation of other products from the reaction mixture in a pure form was not possible. The treatment of the unsymmetrically substituted adduct **3c** with MeSO₃H at 20° for 3 h afforded a complex mixture of products (Scheme 3). A CC of this mixture made it possible to isolate indenofuran **4c** in 25% yield and indenenes **6a**, **6b** in 30% overall yield as inseparable mixtures of isomers. By increasing the reaction time to 3 d, isomeric indenenes **6a**, **6b** were isolated in 47% overall yield, indenofuran **4c** (7%) and a mixture of unidentified products. The reaction of indenofuran **4c** with MeSO₃H in CH₂Cl₂ for 2 d led to indene **6a** in 34% yield (Scheme 4). The structure of **4c** was established unequivocally by X-ray diffraction analysis (Fig. 2). Single crystals of **4c** were obtained by crystallization from CH₂Cl₂/MeOH.

Based on the above results, a plausible mechanism for the formation of compounds **4a** – **4c** is proposed in Scheme 5. The acid-catalyzed isomerization of **3** starts from cation **7**, which undergoes rearrangement through a cascade of cations **8** – **10**. The breaking up of an oxygen bridge in cation **7** gives cyclopropyl carbinyl cation **8**. Subsequent rearrangement leads to homoallylic cation **9**. Finally, a ν [$\pi 2 + \sigma 2 + \sigma 2$]-rearrangement occurs in the

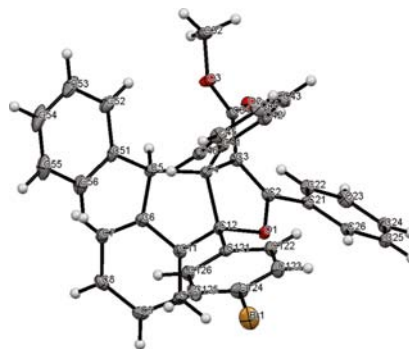
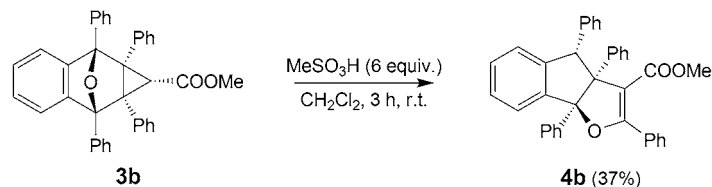


Fig. 2. Molecular structure (ORTEP 50% probability level) of **4c**. Selected interatomic distances [Å] and angles [°]: O(1)–C(2) 1.3682 (18), O(1)–C(12) 1.4597(17), C(3)–C(2) 1.351(2), C(3)–C(4) 1.531(2), C(4)–C(5) 1.595(2), C(6)–C(5) 1.509(2), C(11)–C(6) 1.382(2), C(4)–C(12) 1.592(2), C(2)–O(1)–C(12) 107.97(11), O(1)–C(12)–C(4) 104.67(11), C(2)–C(3)–C(4) 110.57(13), C(3)–C(4)–C(5) 107.79(12), C(3)–C(4)–C(12) 98.35(11).

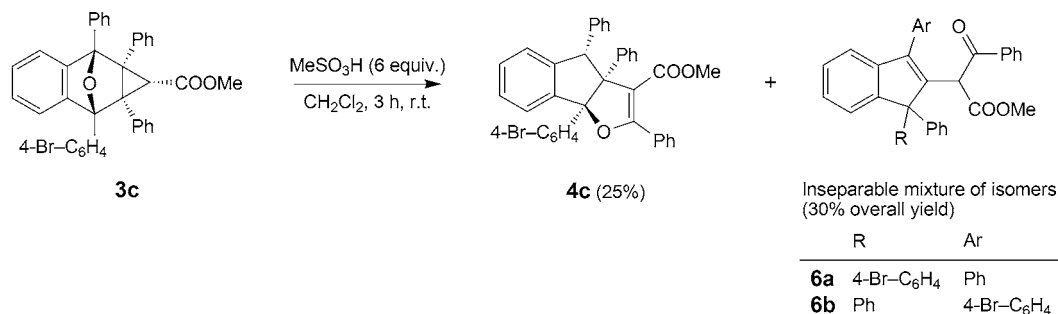
cation **9** to form cation **10**, which *via* deprotonation gives products **4a** – **4c**.

The probable mechanism of formation of indenenes **5** and **6** is shown in Scheme 6. On the first stage, indenofuran **4** reacts with MeSO₃H giving the protonated form **11**,

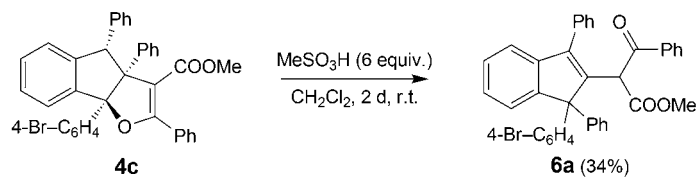
Scheme 2. Reaction of Diels–Alder adduct **3b** with MeSO₃H.

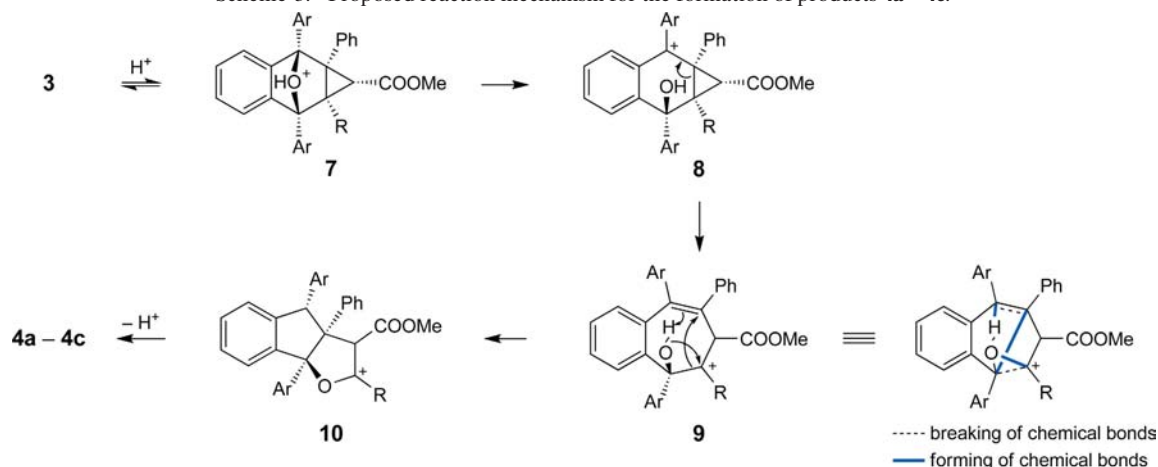


Scheme 3. Reaction of Diels–Alder adduct **3c** with MeSO₃H.

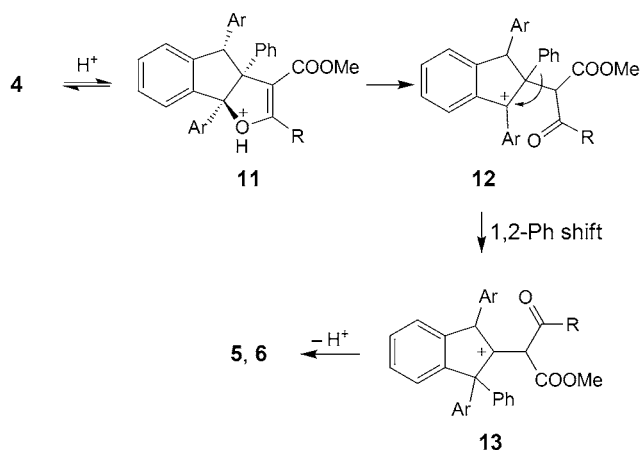
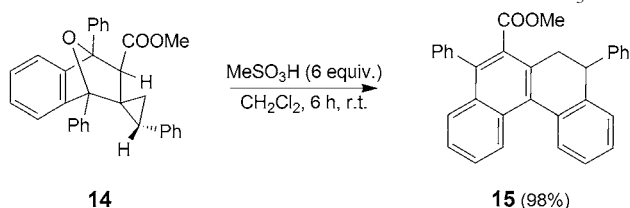


Scheme 4. Reaction of indenofuran adduct **4c** with MeSO₃H.



Scheme 5. Proposed reaction mechanism for the formation of products **4a–4c**.

Scheme 6. Proposed reaction mechanism for the formation of inden-5 and 6.

Scheme 7. Reaction of *Diels–Alder* adduct **14** with $MeSO_3H$.

which is converted into cation **12** through the opening of the furan ring. The latter undergoes a 1,2-phenyl shift with formation of intermediate **13**, which is further transformed into products **5** and **6**.

The choice of spiro-adduct **14** as starting material in this reaction appeared particularly interesting in order to expand the synthetic utility of this methodology (Scheme 7). An adduct **14** was prepared by heating of a mixture of methyl (*E*)-2-(2-phenylcyclopropylidene)acetate and isobenzofuran **2a** in toluene (3 h, 110°). Synthesis, and spectroscopic, analytical, and X-ray characterization data for *Diels–Alder* adduct **14** were previously reported

by our group [32]. Compound **14** was reacted with $MeSO_3H$ (6 equiv.) in CH_2Cl_2 at ambient temperature. After stirring for 6 h and extractive workup, the TLC analysis indicated the formation of only one compound. The methyl 5,8-diphenyl-7,8-dihydrobenzo[*c*]phenanthrene-6-carboxylate (**15**) was isolated from the reaction mixture by the preparative TLC in 98% yield. The structure of the benzo[*c*]phenanthrene **15** was characterized by 1H - and ^{13}C -NMR, MS, and IR spectra. The structure of compound **15** was confirmed additionally by the X-ray diffraction analysis (Fig. 3).

A plausible mechanism for the formation of compound **15** is proposed in Scheme 8. The *Diels–Alder* adduct **14**, under acidic conditions, undergoes subsequent conversion into cations **16** and **17**. The latter is converted to the benzyl cation **18** as result a cyclopropane ring opening. Finally, an intramolecular *Friedel–Crafts* reaction occurs in the intermediate **18** to form product **15**.

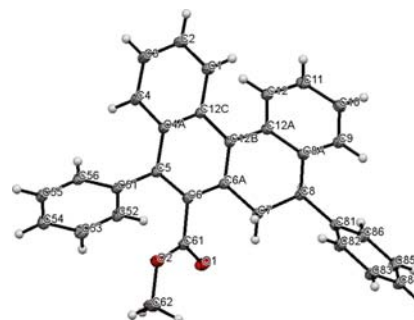
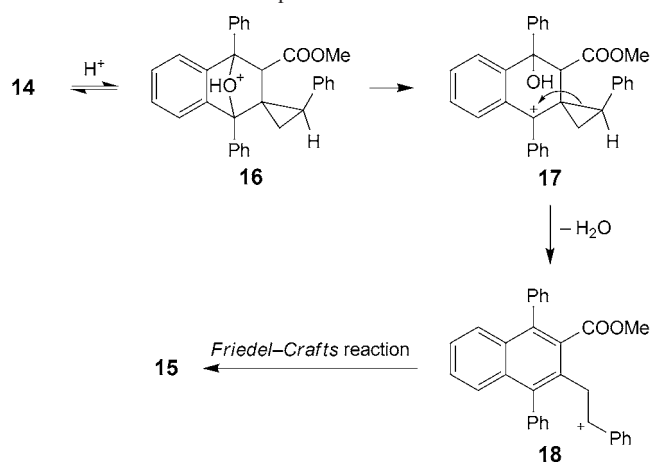


Fig. 3. Molecular structure (ORTEP 50% probability level) of **15**. Selected interatomic distances [Å] and angles [°]: C(6a)–C(7) 1.5142(17), C(6a)–C(12b) 1.3935(17), C(7)–C(8) 1.5393(17), C(8)–C(8a) 1.5236(17), C(8a)–C(12a) 1.4208(17), C(12a)–C(12b) 1.4910(17), C(12b)–C(6a)–C(7) 118.73(10), C(8a)–C(8)–C(7) 107.28(10), C(6a)–C(12b)–C(12a) 117.34(11), C(12)–C(12a)–C(12b) 123.15(11), C(6)–C(6a)–C(7) 120.97(11).

Scheme 8. Proposed reaction mechanism for the formation of product **15**.

Conclusion

We have shown that treatment of the *Diels–Alder* adducts of cyclopropenecarboxylates and 1,3-diarylisobenzofurans with a strong acid triggers a skeletal rearrangement resulting in 4,8b-dihydro-3a*H*-indeno[1,2-*b*]furans. Under similar conditions, the *Diels–Alder* adduct of methyl (*E*)-2-(2-phenylcyclopropylidene)acetate and 1,3-diphenylisobenzofuran is converted to benzo[*c*]phenanthrene derivative.

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Experimental Part

General

Hexane and AcOEt for the chromatography were distilled before use. Cyclopropenes and 1,3-diarylisobenzofurans were prepared following known procedures [33–36]. Reactions were monitored by the TLC analysis using *Silufol* UV-254 plates. The CC was performed on silica gel (SiO_2 ; 70–230 mesh; *Merck*), eluted with AcOEt/hexane. M.p.: *Boetius* instrument; uncorrected. IR Spectra: *Bruker Tensor 27* spectrophotometer. 1H - and ^{13}C -NMR spectra: in $CDCl_3$ using a *Bruker Avance 400* or a *Bruker DPX-300* spectrometer. HR-ESI-MS: *Bruker-microTOF* and *Bruker-maXis* (QTOF) instrument. The X-ray diffraction data were performed by a *Bruker APEX-II* CCD diffractometer with MoK_α X-ray radiation.

General Procedure for the Synthesis of Diels–Alder Adducts **3b** and **3c**

A soln. containing 1 mmol of cyclopropenecarboxylate **1b** and 1.2 mmol 1,3-diarylisobenzofuran **2a**, **2b** in 5 ml of dry benzene was heated at 60° for 3 h. The mixture was concentrated under reduced pressure and the residue was subjected to SiO_2 chromatography using a hexane/AcOEt mixture as eluent.

The reaction of *Diels–Alder* adducts **3a**–**3c** and **14** with $MeSO_3H$ (*General Procedure*). $MeSO_3H$ (6 equiv.) was added at r.t. to a soln. of **3** (1 equiv.) in CH_2Cl_2 (5 ml). After 3 h (TLC-control), 5 ml of Na_2CO_3 (aq.) were added carefully to the mixture. The aq. layer was extracted with CH_2Cl_2 (3×5 ml), the org. layers were combined, dried ($MgSO_4$), and evaporated to dryness. The residue was then purified by a CC using a mixture of hexane/AcOEt as eluent.

Methyl (1*RS*,1*aRS*,2*SR*,7*RS*,7*aRS*)-1a,2,7,7a-Tetrahydro-1a-methyl-2,7,7a-triphenyl-1*H*-2,7-epoxycyclopropa[*b*]naphthalene-1-carboxylate (3a). A soln. containing 3.0 mmol of cyclopropenecarboxylate **1a** and 3.0 mmol 1,3-diphenylisobenzofuran (**2a**) in 50 ml of dry benzene was heated at reflux for 1 h. The mixture was concentrated under reduced pressure and the residue was subjected to SiO_2 chromatography (using hexane/AcOEt mixture as the eluent) to give the product **3a**. Yield: 1.17 g (83%). Colorless crystals. M.p. $120–122^\circ C$. IR ($CHCl_3$): 1050, 1085, 1150, 1175, 1280, 1320, 1345, 1400, 1450, 1490, 1710, 2860, 3020. 1H -NMR (400 MHz, $CDCl_3$): 1.38 (s, Me); 3.60 (br. s, COOMe + $CH_{cycloprop.}$); 6.28 (d, $J = 8$, 2 $H_{arom.}$); 7.16–7.40 (m, 11 $H_{arom.}$); 7.50–7.60 (m, 4 $H_{arom.}$); 7.85 (d, $J = 8$, 2 $H_{arom.}$). ^{13}C -NMR (100 MHz, $CDCl_3$): 10.0 (Me); 32.1 (CH); 41.4 ($C_{cycloprop.}$); 51.0 ($C_{cycloprop.}$); 51.5 (COOMe); 90.5 (C–O); 90.7 (C–O); 122.1; 122.2; 126.0; 126.5; 126.8 (2 C); 127.4 (2 C); 127.5; 127.86; 127.94 (2 C); 128.4; 128.8 (2 C); 128.9 (2 C); 129.0; 132.0 (2 C); 134.3; 134.8; 147.5; 147.9; 169.9 (CO). HR-ESI-MS: 481.1774 ($C_{32}H_{26}NaO_3^+$, $[M + Na]^+$; calc. 481.1780).

Methyl (1*aRS*,2*RS*,7*SR*,7*aSR*)-1a,2,7,7a-Tetrahydro-1a,2,7,7a-tetraphenyl-1*H*-2,7-epoxycyclopropa[*b*]naphthalene-1-carboxylate (3b). Yield: 469 mg (80%). Colorless crystals. M.p. $265–267^\circ C$. IR ($CHCl_3$): 1175, 1195, 1303, 1330, 1447, 1496, 1737, 2950, 3053. 1H -NMR (400 MHz, $CDCl_3$): 3.33 (s, COOMe); 3.83 (s, $CH_{cycloprop.}$); 6.67 (d, $J = 8$, 4 $H_{arom.}$); 7.13–7.40 (m, 20 $H_{arom.}$). ^{13}C -NMR (100 MHz, $CDCl_3$): 32.7 ($CH_{cycloprop.}$); 51.1 (COOMe); 54.8 (2 $C_{cycloprop.}$); 90.2 (2 C^*O); 122.5 (2 C); 126.3 (2 C); 126.7 (4 C); 127.2 (4 C); 127.3 (2 C); 127.89 (4 C); 127.92 (2 C); 131.9 (2 C); 132.2 (4 C); 134.6 (2 C); 147.5 (2 C); 169.3 (CO). HR-ESI-MS: 543.1931 ($C_{37}H_{28}NaO_3^+$, $[M + Na]^+$; calc. 543.1936).

Methyl (1*SR*,1*aSR*,2*SR*,7*RS*,7*aRS*)-2-(4-Bromophenyl)-1a,2,7,7a-tetrahydro-1a,7,7a-triphenyl-1*H*-2,7-epoxycyclopropa[*b*]naphthalene-1-carboxylate (3c). Yield: 310 mg (52%). Colorless crystals. M.p. $209–211^\circ C$. IR ($CHCl_3$):

1009, 1168, 1331, 1397, 1445, 1493, 1602, 1745, 2948, 3053. $^1\text{H-NMR}$ (400 MHz, CDCl_3): 3.34 (s, COOMe); 3.80 (s, $\text{CH}_{\text{cycloprop.}}$); 6.65 (d, $J = 7$, 4 $\text{H}_{\text{arom.}}$); 7.16–7.40 (m, 17 $\text{H}_{\text{arom.}}$); 7.50 (d, $J = 8$, 2 $\text{H}_{\text{arom.}}$). $^{13}\text{C-NMR}$ (100 MHz, CDCl_3): 32.5 ($\text{CH}_{\text{cycloprop.}}$); 51.2 (COOMe); 54.8 (2 $\text{C}_{\text{cycloprop.}}$); 90.0 (C–O); 90.6 (C–O); 122.1; 122.3; 122.6; 126.3; 126.4; 126.6 (2 C); 127.2 (2 C); 127.3 (2 C); 127.4; 127.5; 127.9 (2 C); 128.0; 128.3 (2 C); 131.1 (2 C); 131.6; 131.7; 132.1 (4 C); 133.5; 134.3; 147.1; 147.4; 169.1 (CO). HR-ESI-MS: 621.1036 ($\text{C}_{37}\text{H}_{27}^{79}\text{BrNaO}_3^+$, $[M + \text{Na}]^+$; calc. 621.1041), 623.1015 ($\text{C}_{37}\text{H}_{27}^{81}\text{BrNaO}_3^+$, $[M + \text{Na}]^+$; calc. 623.1015).

Methyl (3aSR,4RS,8bRS)-4,8b-Dihydro-2-methyl-3a,4,8b-triphenyl-3aH-indeno[1,2-b]furan-3-carboxylate (4a) and **Methyl 3-Oxo-2-(1,1,3-triphenyl-1H-inden-2-yl)butanoate (5)**. Following the *General Procedure*, Diels–Alder adduct **3a** (0.76 mmol) and MeSO_3H (4.5 mmol) were reacted in CH_2Cl_2 . CC gave **4a** (121 mg, 35%) and **5** (51 mg, 15%).

Data of **4a**: Colorless crystals. M.p. 198 – 200 °C. IR (CHCl_3): 1096, 1131, 1193, 1243, 1375, 1446, 1496, 1634, 1683, 3053. $^1\text{H-NMR}$ (300 MHz, CDCl_3): 2.41 (s, Me); 3.88 (s, COOMe); 5.67 (s, CH); 6.05 (d, $J = 6$, 2 $\text{H}_{\text{arom.}}$); 6.56 – 6.80 (m, 5 $\text{H}_{\text{arom.}}$); 6.90 – 7.15 (m, 8 $\text{H}_{\text{arom.}}$); 7.38 – 7.55 (m, 4 $\text{H}_{\text{arom.}}$). $^{13}\text{C-NMR}$ (100 MHz, CDCl_3): 16.1 (Me); 51.3 (COOMe); 60.0 (CH); 71.2; 106.3; 110.3; 125.3; 125.8 (2 C); 126.8; 127.0 (3 C); 127.2 (2 C); 127.6; 127.9 (2 C); 128.5 (3 C); 130.1; 130.5; 130.6 (2 C); 138.7; 140.5; 143.1; 144.0; 146.1; 167.5 (=C–O); 168.7 (CO). HR-ESI-MS: 481.1774 ($\text{C}_{32}\text{H}_{26}\text{NaO}_3^+$, $[M + \text{Na}]^+$; calc. 481.1780).

Crystallographic Data for **4a**. $\text{C}_{32}\text{H}_{26}\text{O}_3$, M_r 458.53, monoclinic, $P2_1/c$ (no. 14), $a = 9.76434(9)$, $b = 9.59139(11)$, $c = 25.5371(3)$ Å, $\beta = 95.6193(9)^\circ$, $V = 2380.15(4)$ Å³, $Z = 4$, $D_x = 1.280$ g cm^{−3}, $F(000) = 968$, radiation, $\text{CuK}\alpha$ ($\lambda = 1.54184$ Å), $3.48 \leq 2\theta \leq 76.60$, intensity data were collected at 293(2) K with a Bruker APEX-II CCD diffractometer, and by employing a $w/2\theta$ scanning technique, in the range of $-12 \leq h \leq 1$, $-12 \leq k \leq 12$, $-32 \leq l \leq 29$; the structure was solved by a direct method; all non-H-atoms were positioned, and anisotropic thermal parameters were refined from 5001 observed reflections with $R(\text{into}) = 0.0140$ by a full-matrix least-squares technique converged to $R = 0.0371$ and $R_w = 0.0925$ [$I > 2\sigma(I)$]. CCDC No.: 1057280.

Data of **5**: White crystals. M.p. 182 – 184 °C. IR (CHCl_3): 1032, 1043, 1070, 1090, 1118, 1154, 1193, 1230, 1246, 1346, 1432, 1445, 1479, 1491, 1596, 1726, 3056. $^1\text{H-NMR}$ (400 MHz, CDCl_3): 2.24 (s, Me); 3.75 (s, COOMe); 4.07 (s, CH); 6.59 – 7.40 (m, 19 $\text{H}_{\text{arom.}}$). $^{13}\text{C-NMR}$ (100 MHz, CDCl_3): 31.0 (Me); 52.0 (COOMe); 54.6 (CH); 64.5; 125.8; 126.5; 126.8; 127.5 (2 C); 127.6 (2 C); 127.8 (2 C); 127.9 (3 C); 128.0 (2 C); 128.4 (2 C); 128.6 (2 C); 130.4; 133.3; 135.1; 137.9; 138.7; 139.0; 140.7; 141.3; 172.2 (CO); 208.2 (CO). HR-ESI-MS: 481.1774 ($\text{C}_{32}\text{H}_{26}\text{NaO}_3^+$, $[M + \text{Na}]^+$; calc. 481.1780).

Methyl (3aS,4R,8bR)-4,8b-Dihydro-2,3a,4,8b-tetraphenyl-3aH-indeno[1,2-b]furan-3-carboxylate (4b). 29 mg (37%) was obtained using *General Procedure* from Diels–Alder adduct **3b** (77 mg). White crystals. M.p. 225 – 227 °C. IR

(CHCl_3): 1075, 1180, 1330, 1420, 1470, 1620, 1720, 3055. $^1\text{H-NMR}$ (400 MHz, CDCl_3): 3.75 (s, COOMe); 5.67 (s, CH); 6.19 (d, $J = 8$, 2 $\text{H}_{\text{arom.}}$); 6.64 (t, $J = 8$, 2 $\text{H}_{\text{arom.}}$); 6.78 – 6.85 (m, 3 $\text{H}_{\text{arom.}}$); 6.95 – 7.10 (m, 5 $\text{H}_{\text{arom.}}$); 7.12 – 7.14 (m, 3 $\text{H}_{\text{arom.}}$); 7.40 – 7.47 (m, 4 $\text{H}_{\text{arom.}}$); 7.50 – 7.67 (m, 3 $\text{H}_{\text{arom.}}$); 7.69 (d, $J = 8$, 2 $\text{H}_{\text{arom.}}$). $^{13}\text{C-NMR}$ (100 MHz, CDCl_3): 51.0 (COOMe); 59.5 (CH); 72.3; 105.6; 110.3; 125.3; 125.6 (2 C); 126.5 (2 C); 126.7; 127.0 (2 C); 127.5; 127.6 (2 C); 127.8 (2 C); 128.3; 128.4 (2 C); 129.4 (2 C); 129.9; 130.36; 130.41 (2 C); 130.6 (2 C); 130.9; 138.3; 140.0; 142.5; 143.5; 145.6; 165.5 (=C–O); 166.2 (CO). HR-ESI-MS: 543.1931 ($\text{C}_{37}\text{H}_{28}\text{NaO}_3^+$, $[M + \text{Na}]^+$; calc. 543.1936).

Methyl (3aSR,4RS,8bRS)-8b-(4-Bromophenyl)-4,8b-dihydro-2,3a,4-triphenyl-3aH-indeno[1,2-b]furan-3-carboxylate (4c), **Methyl 2-[1-(4-Bromophenyl)-1,3-diphenyl-1H-inden-2-yl]-3-oxo-3-phenylpropanoate (6a)**, and **Methyl 2-[3-(4-Bromophenyl)-1,1-diphenyl-1H-inden-2-yl]-3-oxo-3-phenylpropanoate (6b)**. Following the *General Procedure*, Diels–Alder adduct **3c** (0.5 mmol) and MeSO_3H (3.0 mmol) were reacted in CH_2Cl_2 . The column chromatography gave **4c** (75 mg, 25%) and **6a/6b** (91 mg, 30% overall yield).

MeSO_3H (6 equiv.) was added at r.t. to a soln. of adduct **4c** (1 equiv.) in CH_2Cl_2 (7 ml). After 2 d of the reaction (TLC-control), the 7 ml of Na_2CO_3 (aq.) was added carefully to the mixture. The aq. layer was extracted with CH_2Cl_2 (3×5 ml), the org. layers were combined, dried over MgSO_4 , and evaporated to dryness. The residue was then purified by a CC using a mixture of hexane/AcOEt as eluent. The residue was subjected to SiO_2 CC (using hexane/AcOEt mixture as the eluent) to give the product **6a** in 34% yield.

Data of **4c**: White crystals. M.p. 177 – 179 °C. IR (CHCl_3): 1101, 1155, 1187, 1251, 1283, 1327, 1395, 1444, 1489, 1572, 1596, 1610, 1704, 3053. $^1\text{H-NMR}$ (400 MHz, CDCl_3): 3.75 (s, COOMe); 5.67 (s, CH); 6.19 (d, $J = 8$, 2 $\text{H}_{\text{arom.}}$); 6.65 – 7.70 (m, 21 $\text{H}_{\text{arom.}}$). $^{13}\text{C-NMR}$ (100 MHz, CDCl_3): 51.0 (COOMe); 59.4 (CH); 72.3; 105.0; 110.5; 125.6; 125.9 (2 C); 126.3; 126.7; 126.8; 127.66; 127.70 (2 C); 127.8 (2 C); 128.5; 129.4 (3 C); 130.1 (2 C); 130.2 (2 C); 130.3 (2 C); 130.5; 130.6 (2 C); 130.7; 138.0; 139.2; 142.3; 143.0; 145.6; 165.3 (=C–O); 206.0 (CO). HR-ESI-MS: 621.1036 ($\text{C}_{37}\text{H}_{27}^{79}\text{BrNaO}_3^+$, $[M + \text{Na}]^+$; calc. 621.1041), 623.1021 ($\text{C}_{37}\text{H}_{27}^{81}\text{BrNaO}_3^+$, $[M + \text{Na}]^+$; calc. 623.1015).

Crystallographic Data for **4c**. $\text{C}_{37}\text{H}_{27}\text{BrO}_3$, M_r 599.50, triclinic, $P\bar{1}$ (no. 2), $a = 9.2308(3)$, $b = 12.6850(4)$, $c = 13.0536(5)$ Å, $\alpha = 86.634(3)^\circ$, $\beta = 81.895(3)^\circ$, $\gamma = 71.113(3)^\circ$, $V = 1431.60(8)$ Å³, $Z = 2$, $D_x = 1.391$ g cm^{−3}, $F(000) = 616$, radiation, $\text{CuK}\alpha$ ($\lambda = 1.54184$ Å), $3.65 \leq 2\theta \leq 76.49$, intensity data were collected at 100(2) K with a Bruker APEX-II CCD diffractometer, and employing $w/2\theta$ scanning technique, in the range of $-9 \leq h \leq 11$, $-15 \leq k \leq 15$, $-15 \leq l \leq 15$; the structure was solved by a direct method; all non-H-atoms were positioned, and anisotropic thermal parameters were refined from 5375 observed reflections with $R(\text{into}) = 0.0194$ by a full-matrix least-squares technique converged to $R = 0.0307$ and $R_w = 0.0703$ [$I > 2\sigma(I)$]. CCDC-1057421.

Data of **6a**: White solid. M.p. 232 – 233 °C. IR (CHCl₃): 1035, 1070, 1195, 1275, 1400, 1435, 1470, 1600, 1725, 3060. ¹H-NMR (400 MHz, CDCl₃): 3.40 (s, COOMe); 4.86 (s, CH); 6.34 – 7.75 (m, 23 H_{arom.}). ¹³C-NMR (100 MHz, CDCl₃): 67.3; 72.7; 121.4 (2 C); 126.9 (2 C); 127.1; 127.3 (2 C); 127.6 (2 C); 127.7; 127.9; 128.7 (2 C); 129.5 (2 C); 130.2; 130.7 (2 C); 130.8; 131.3; 131.6 (2 C); 131.9; 132.1; 132.5; 132.9; 141.3; 141.5; 142.5; 142.7; 142.9; 143.1; 143.6; 168.9 (CO); 206.6 (CO). HR-ESI-MS: 621.1032 (C₃₇H₂₇⁷⁹BrNaO₃⁺, [M + Na]⁺; calc. 621.1041), 623.1017 (C₃₇H₂₇⁸¹BrNaO₃⁺, [M + Na]⁺; calc. 623.1015).

Data of **6b**: ¹H-NMR (400 MHz, CDCl₃): 3.40 (s, COOMe); 4.86 (s, CH); 6.34 – 7.75 (m, 23 H_{arom.}). ¹³C-NMR (100 MHz, CDCl₃): 52.2; 67.5; 73.1; 121.1 (4 C); 126.9 (2 C); 127.0; 127.1; 127.2; 127.25 (2 C); 127.34 (2 C); 127.6 (2 C); 127.7; 127.9 (2 C); 128.7 (4 C); 129.5 (2 C); 131.1; 131.2; 132.8; 133.0; 140.3; 140.8; 141.2; 143.4; 168.6 (CO); 206.5 (CO). HR-ESI-MS: 621.1036 (C₃₇H₂₇⁷⁹BrNaO₃⁺, [M + Na]⁺; calc. 621.1041). HR-ESI-MS: 623.1015 (C₃₇H₂₇⁸¹BrNaO₃⁺, [M + Na]⁺; calc. 623.1015).

Methyl 5,8-Diphenyl-7,8-dihydrobenzo[c]phenanthrene-6-carboxylate (15). A quantity of 113 mg (98%) was obtained using *General Procedure* (reaction time 6 h) from *Diels–Alder* adduct **14** (120 mg). Yield 98%. White crystals. M.p. 156 – 158 °C. IR (CHCl₃): 1053, 1142, 1201, 1273, 1378, 1436, 1493, 1599, 1725, 3058. ¹H-NMR (300 MHz, CDCl₃): 3.12 – 3.19 (m, 1 H of CH₂); 3.31 – 3.40 (m, 1 H of CH₂); 3.53 (s, COOMe); 4.15 – 4.25 (m, CH); 6.98 (d, *J* = 6, 1 H_{arom.}); 7.25 – 7.60 (m, 15 H_{arom.}); 7.73 (d, *J* = 9, 1 H_{arom.}); 8.00 (d, *J* = 9, 1 H_{arom.}); 8.64 (d, *J* = 9, 1 H_{arom.}). ¹³C-NMR (75 MHz, CDCl₃): 34.9 (CH₂); 45.01 (CH₂); 52.3 (COOMe); 126.1; 126.3; 126.5; 127.3 (2 C); 127.6; 127.8; 127.9; 128.1; 128.4; 128.6; 129.1 (3 C); 129.4 (2 C); 129.6; 130.5; 130.6; 130.8; 131.8; 132.0; 132.6; 133.9; 137.5; 138.5; 142.4; 142.6; 170.2 (CO). HR-ESI-MS: 463.1669 (C₃₂H₂₄NaO₂⁺, [M + Na]⁺; calc. 463.1674).

Crystallographic Data for **15**. C₃₂H₂₄O₂, *M*_r 440.51, triclinic, *P* $\bar{1}$ (no. 2), *a* = 5.7200(2), *b* = 10.3268(4), *c* = 19.9208(8) Å, α = 92.798(3), β = 98.090(3), γ = 94.860(3)°, *V* = 1158.60(8) Å³, *Z* = 2, *D*_x = 1.263 g cm^{−3}, *F*(000) = 464, radiation, MoK α (λ = 0.71073 Å), 2.78 ≤ 2 θ ≤ 31.87, intensity data were collected at 100(2) K with a *Bruker APEX-II* CCD diffractometer, and by employing a *w/2 θ* scanning technique, in the range of $-7 \leq h \leq 8$, $-15 \leq k \leq 15$, $-27 \leq l \leq 28$; the structure was solved by a direct method; all non-H-atoms were positioned, and anisotropic thermal parameters were refined from 7174 observed reflections with *R*(into) = 0.0357 by a full-matrix least-squares technique converged to *R* = 0.0536 and *R*_w = 0.1205 [*I* > 2 σ (*I*)]. CCDC-1057274.

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